

Acidity Function of the Methyl Ether-D<sub>2</sub>O System,  
by G. B. Manelis.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 5,  
1959, pp 1030-1034.

ML 2232

Sci -  
Jul 63

238,182

Self-Diffusion of Dibasic Phosphate Ion in the  
System Water-Electrolyte-Nonelectrolyte, by  
P. Ya. Siver, 9 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 5, 1959,  
pp 1065-1070.

SLA 60-17727

Sci

129, 408

OTS, Vol IV, No 3

Sep 60

The Photocatalytic Properties of Zinc Oxide in  
the Reduction of Methylene Blue. 4. Poisoning  
of the Photocatalysts by Nitro Compounds, by I. N.  
Barshchevskiy.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 5,  
1959, pp 1071-1079.

DSIR LIU RTS 1537

Sci - Chem

125,547

Sep 60

Investigation of the Rate of Burning of Acetylene  
Steel in a Turbulent Flow. FD 2132569

RUDNIAK, per, Zhur Diz Khim, vol XXIX, 1959,  
pp 1060-1095.

Sci. - Phys

Oct 59

3  
ACBx, H-480  
ATS-05-37R  
ATS/RJ-24/73

110,349

<p>Vasil'yev, S. S. KINETIC ANALYSIS OF CHAIN REACTIONS. V. REARRANGEMENT OF THE FORM OF THE SOLU- TIONS OF THE BASIC EQUATIONS OF THE KINET- ICS OF CHAIN REACTIONS FOR THE CASE OF TWO ACTIVE SPECIES (Kineticcheskiy Analiz Tsepmnykh Reaktsii. V. Preobrazovaniye Formy Reshenii Osnovnykh Uravnenii Kinetiki Tsepmnykh Reaktsii dlya Sluchaynykh Dvukh Aktivnykh Chastits). Jan 61 [16]p. 9 refs. RTS 1739. Order from LC or SLA m1\$2.40, ph\$3.30 61-15732  Trans. of Zhur[nnal] Fiz[icheskoy] Khim[ii] (USSR) 1959, v. 33, no. 5, p. 1100-1110.</p> <p>Transformations of the form of the solutions of the basic differential kinetic equations of chain reactions for the case of two active species led to expressions that directly show how the kinetic development of chain reactions is affected by the magnitudes of the generalized rate constants of the initial kinetic (Chemistry--Physical, TT, v. 5, no. 12) (over)</p>	<p>61-15732</p> <p>I. Chemical reactions-- Mathematical analysis 2. Title: Chain reactions I. Vasil'yev, S. S. II. RTS-1739 III. Department of Scientific and Industrial Research (Gt. Brit.) IV. Title: Rearrangement ...</p> <p>100-142</p> <p>Office of Technical Services</p>
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Heat of Adsorption in the Electronic Chemical  
Sorption Theory, by Sh. M. Kogan, V. B.  
Sandomirskiy, 8 pp.

RUSSIAN, Sov. Zhur fiz. chm., Vol. XXIII, 1959,  
pp 1129-1133.

AEC Tr-4899

Sci - Phys, Chem  
21 Feb 61

140,948

Fedoseeva, O. P. and Tunitskii, N. N.  
STUDY OF THE KINETICS OF ION EXCHANGE  
SORPTION. III. COMPLETE EXCHANGE BETWEEN  
MONOVALENT AND DIVALENT ION IN RESINS.  
[1961] [9]p. 4 refs.

Order from OTS or SLA \$1.10

62-10307

Trans. of Zhurnal Fizicheskoi Khimii (USSR) 1959,  
v. 33, [no. 5] p. 1140-1146.

DESCRIPTORS: \*Ion exchange resins, Resins, Hydrogen, Sodium, Calcium, Strontium, Ions, Exchange reactions, Reaction kinetics, Diffusion, Adsorption.

The effect was studied of the cationic form of the resin on the diffusion coefficients of ionic microcomponents. It is shown that monovalent as well as divalent ions diffuse faster in a resin containing monovalent ions than in a resin containing divalent ions. Swelling of (Chemistry--Physical, TF, v. 7, no. 10) (over)

62-10307

I. Fedoseeva, O. P.  
II. Tunitskii, N. N.  
III. Title: Complete ...

Office of Technical Services

# 19

(DC-3692)

Boris Vasil'yevich Yerofeyev (On his 50th  
Birthday), by S. V. Markevich, 4 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 5,  
1959, pp 1155, 1156.

JPRS 3368

USSR  
Biog  
Jun 60

117,571

Study of the Oxidation of Cyclohexane by  
the Method of Inhibition During the Reaction,  
by E. T. Denisov, 13pp.  
RUSSIAN, per, Zhurnal Fizicheskoi Khimii,  
Vol 33, 1959, No 6, pp 1198-1208.  
CFSTI TT-64-18412

323,208

Sci - Chemistry  
Apr 67

Presentation of the Equation of State of Real  
Gas in Explicit Form in Terms of the Independent  
Variables T and v, by V. A. Zagoruchenko,  
P. M. Kassel'man, 8 pp.  
RUSSIAN, per, Zhurnal Fizicheskoy Khimii, Vol 33,  
No 6, 1959, pp 1221-1229.  
NTIS ID: TT 69-55051

Sci/Chem  
Dec 70

Investigation of the Surface State of Nickel  
and Iron Cathodes From the Drop in Potential  
After Disconnecting the Current, by

V. E. Past, Z. A. Iofa.

RUSSIAN, per, Zhurnal Fizicheskoy Khimii,

Vol 33, No 6, 1959, pp 1230-1237.

NASA TT Full, 770

Sci  
Dec 68

371,123

A Study of the Wetting of Solid Surfaces of Refractory Oxides by Liquid Metals, by V. N. Yeremenko,  
Yu. V. Maydich.  
RUSSIAN, per, Zhurnal Fiz Khimii, Vol 33, No 6,  
1959, pp 1238-1245.  
CIA X-6779

Oct 68

369,408

Investigation of the Influence of the Structure of Unsaturated Compounds on Their Reactivity in Copolymerization Processes. Part 2. The Effect of Different Substituents Attached to the Nitrogen on the Reactivity of Methacrylamide Derivatives With Respect to a Common Type of Radical, by G. A. Petrova, G. A. Shtreykhman, 9 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 6, 1959, pp 1245-1252.

195, 969

Sci

Vol 4, No 10, May 62

AT&T-2344R  
AT&T-2450

The Effect of Ionizing Radiation on the Corrosion of  
Metals Under Atmospheric Conditions, 11 pp.

UD 67623902  
RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 6,  
1959, PP 1256-1262.

ACSI, H-4619

Sci - Chem

TIC/CI-4 T-2171

Nov 59

100,509

<p>Zosimovich, D. P. and Bogatova, N. F. ELECTRO DEPOSITION OF ZINC IN THE PRESENCE OF SMALL QUANTITIES OF ANTIMONY AND COBALT (Elektrolyticheskoye Vydeleniye Tsinka v Prisutstvii Malykh Kolichestv Suri'my i Kobalta). Nov 60 [8]p. 10 refs. RTS 1692. Order from LC or SLA m\$1.80, ph\$1.80 61-15219  Trans. of Zhurnal Fizicheskoy Khimii (USSR) 1959, v. 33, no. 6, p. 1324-1327.</p> <p>It is established that in the presence of small quantities of antimony (0.05 mg/l) and cobalt (20 mg/l) during the electrolytic deposition of zinc an increase in polarisation is observed. When cobalt is present in solutions containing 0.2 mg/l of antimony, the process is determined by the presence of antimony. Antimony, introduced into the electrolyte either as potassium antimonyl tartrate (tartar emetic) or as antimony sulphate, has the same influence on the course of the polarisation curves. It is suggested that small (Chemistry--Physical, TT, v. 5, no. 7) (over)</p>	<p>61-15219</p> <p>I. Zinc--Electrodeposition I. Zosimovich, D. P. II. Bogatova, N. F. III. RTS-1692 IV. Department of Scientific and Industrial Research (Gt. Brit.)</p> <p><i>148,730</i></p>
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Office of Technical Services

Phase Equilibria in the System Melamine-Ammonia,  
by I. P. Krichevskiy, G. D. Efremova,

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 6,  
1959, pp 1328-1335.

ATS RJ-2040

Sci - Chem  
Dec 59

103,974

A Method for Calculating Vapor-Liquid Equilibria  
of Tricomponent Systems, by V. B. Kogan, V. M.  
Safronov, 11 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, No 6,  
1959, pp 1353-1359.

SLA 60-17729

Sci

129, 401

OTS, Vol IV, No 3

Sep 60

<p>Shilov, A. Ye. and Sabirova, R. D. MECHANISM OF THE PRIMARY ACT OF DECOMPOSITION OF CHLORINE DERIVATIVES OF METHANE. I. DECOMPOSITION OF CARBON TETRACHLORIDE AND METHYL CHLORIDE (Mekhanizm Pervichnogo Akta Termicheskogo Razpada Khloroprolizvodnykh Metana) Nov 60 [17]p. 19 ref. RTS 1527. Order from LC or SLA m\$2.40, ph\$3.30 t1-13488 Trans. of Zhurnal Priborostroyeniya (USSR) 1959 [v. 33] no. 6, p. 1345-1373.</p> <p>The kinetics of the thermal decomposition of carbon tetrachloride and methyl chloride in jets was investigated. The primary decomposition in both cases takes place with the disruption of a chlorine atom. A mechanism for the secondary reaction is suggested, which agrees with the products of reaction and the kinetic data. (Author)</p> <p>(Chemistry--Physical, TT, v. 5, no. 4)</p>	<p>61-13488</p> <p>I. Chlorides--Decomposition I. Shilov, A. Ye. II. Sabirova, R. D. III. Title: Decomposition... IV. RTS-1527 V. Department of Scientific and Industrial Research (Gt. Brit.)</p> <p>143,240</p> <p>Office of Technical Services</p>	
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<p>Kazanskii, K. S., Entelis, S. G., and Chirkov, N. M. THE SOLUBILITY OF GASEOUS ISOBUTYLENE IN WATER. [1962] 6p. Order from ATS \$8.95</p> <p>Trans. of Zhurnal Priborostroyeniia i Khimii (USSR) 1959, v. 33, no. 6, p. 1409-1413.</p> <p>DESCRIPTORS: *Butenes, Gases, Solubility, Water <i>ML 822 Nov (110)</i></p> <p>(Chemistry--Physical, TT, v. 9, no. 2)</p>	<p>63-12262</p> <p>I. Kazanskii, K. S. II. Entelis, S. G. III. Chirkov, N. M. IV. ATS-97P64R V. Associated Technical Services, Inc., East Orange, N. J.</p> <p>ATS KJ-2942</p> <p><i>Murphy</i> <i>Canned in</i> <i>NATE 6-2778</i></p> <p>Office of Technical Services</p>
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Absorption Properties of Cellulose in the Process of  
Its Thermal Decomposition, by T. V. Gatovskaya.  
RUSSIAN, per, Zhur Fiz Khim., Vol 33, No 6, 1959,  
pp 1418-1421.  
\*NTIS TT 71-51001

July 71

Marinin, V.A.  
VARIATION THERMIQUE DU COEFFICIENT DE DIFFUSION DE CERTAINES SUBSTANCES DANS LE TÉTRACHLORÉTHANE ET LE TÉTRABROMÉTHANE  
(Temperature Relation of the Diffusion Coefficient of Substances in Tetrachloro - and Tetra bromoethane).  
5p. 5refs. CNRS-VIII bis 690  
Order from OTS, ETC or CNRS \$0.80 TT-62-26386

Trans. in French of Zhurnal Fizicheskoi Khimii  
(USSR) 1956, v. 33, no. 6, p.1430-1433.

DESCRIPTORS: Bromine compounds, Chlorine compounds, Hydrogen compounds, Carbon compounds,  
\*Diffusion, Heat, \*Ethyanes.

(Chemistry--Physical, TT, v. 11, no. 7)

TT-62-26386

I. Marinin, V.A.  
II. CNRS-VIII bis 690  
III. Centre National de la Recherche Scientifique, Paris

Office of Technical Services  
European Translations Centre

The Thermodynamics of the Superstate, by V. K.  
Semenchenko, 8 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXIII, 1959,  
pp 1440-1445.

ABC Tr-4271

Sci - Phys  
9 Mar 61

141,572

Applying Thermodynamics of Irreversible Processes  
to the Phase Theory. The Case of the Two Component  
Two-Phase System at Constant Temperature and  
Pressure, by Kiril A. Popov.  
RUSSIAN, per, Zh. Fiz. Khim., Vol 32, 1958,  
pp 1638-47.  
WAPD-tr-S7

Sci - Chem  
Oct 67

342,701

Ion Hydration and Its Mechanism. The Structure  
and Dimensions of Hydrated Ions, by A. F.  
Kapustinskii, 15 pp.

RUSSIAN, per, Zhur Fiz Khim, 1958, Vol XXXII,  
No 7, pp 1648-1657.

SLA 60-17725

Sci

3 Mar 62

186, 171

Andreeva, V. V. and Shishakov, N. A.  
DE L'ÉPAISSEUR DES PELLICULES D'OXYDES SUR  
CERTAINES MÉTALLUX SELON LES DONNÉES DES  
ÉTUDES ELECTRONOGRAPHIQUES ET OPTIQUES  
[On the Thickness of Oxide Films on Some Metals from  
the Data of Electron Diffraction and Optical Investiga-  
tions] tr. by de Trezvinsky, 22 Mar 60 [8]p. 5 refs.  
CEA Trans. no. R 855 (text in French).  
Order from OTS or SLA \$1.10 61-23816

Trans. in French of Zhurnal Piz[icheeskoi] Khim[II]  
(USSR) 1958, v. 32, no. 7, p. 1671-1672.  
A translation in English is available from IC or SLA  
m\$1.80, p\$1.80 as ABC-tr-3955, Jan 60, 5p.

DESCRIPTORS: \*Metals. \*Oxidation. \*Thin films.  
\*Oxides. Electron diffraction analysis. Optical analysis.  
Thickness.

For abstract see Technical Translations 3: 776, 1960.  
(Chemistry--Physical, TT, v. 6, no. 6)

61-23816

I. Andreeva, V. V.  
II. Shishakov, N. A.  
III. CEA-tr-R855  
IV. Commissariat à l'Énergie  
Atomique (France)

176696

Office of Technical Services

P- cat copy Arsenal  
Req. 154-71

R-13363-B  
28 June 1971

Zhur. Fiz. Khim. Vol. 32, pp. 1690-1691. (1958).  
(Russian Journal of Physical Chemistry)

Medvedev, V. A.

Language: Russian

Please translate & type 1 copy only.

The Approximation Computation of the Temperature  
Function of Heat Capacity, by M. Kh. Karapet'-  
yants, 14 pp.

RUSSIAN, per, Zhar Fiz Khim, Vol XXXIII, No 8,  
1953, pp 1763-1773.

OTS 60-17719

Sci

188, 910

30 Mar 62

Kinetics of Catalytic Cracking of Petroleum,  
by G. N. Panahenkow, A. S. Matanikaya, 6 pp.

REVIEW, par, Thor Pin Khin (REB), Vol 2222,  
No 8, 1956, pp 1779-1783.

Asiac Tech Serv

571208

ATS/RJ-1595

Set .. Chem  
May 59  
CMB, v 3, No 6

86,347

The Mechanism of the Cathode Process in the  
Electrolysis of a Fused NaCl.-CaCl<sub>2</sub> Mixture on  
an Iron Cathode, by L. Suski, 22<sup>2</sup> 19 pp.

NUSSLIAN, per, Zhur Fiz Khim, Vol XXXII, 1958,  
pp 1785-1795.

AEC Tr 3658

Sci - Phys

Dept of Defense  
7/7/58 No 303

Jun 59

88,899

Catalytic Properties of Cerium Dioxide in the  
Dehydrogenation and Dehydration of Alcohols  
and in the Dehydrogenation of Cyclohexane, by  
A. A. Tolstopyatova, A. A. Balandin.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 8,  
1958, pp 1831-1841.

ATS RJ-1903

11/6/1958

Sci - Chem  
Dec 59

104/080

Antonovskaya, E.I. and Sukhotin, A.M.  
SUR LES PARTICULARITES DE L'ACTIVATION  
ANODEIQUE DU CHROME ET DES ACIERS AU CHROME  
II (Ob Osobennostyakh Anodnogo Aktivirovaniya Khroma  
i Khromistykh Stalei). II (Familiarities of the Anodic  
Activation of Chromium and Chromium Steels. II) 14p.  
(Foreign text include 9 5refs., CNRS-VIII bis 584,  
Order from OIS, ETC or CNRS 50,80 TT-62-28263

Trans. in French of *Zhurnal Fizicheskoi Khimii* (USSR)  
1958, v. 32, no. 8, p. 1442-1450.  
Trans. in English available from ATS \$8.00 as  
ATS-6NM4IR.

DESCRIPTIONS: \*Chromium, \*Chromium alloys,  
Iron alloys, Steel, Anodes, Activity coefficient, pH,

(Metallurgy--Ferrous Metals, TF, v. 11, no. 8)

TT-62-28263

I. Antonovskaya, E.I.  
II. Sukhotin, A.M.  
III. CNRS-VIII bis 584  
IV. Centre National de la Recherche Scientifique, Paris

Office of Technical Services  
European Translations Centre

S-322/60

(NY-3919)

Studies of Thermodynamic Properties of Substances  
With the Aid of the Method of Explosion in a  
Spherical Bomb, by V. A. Medvedev, 12 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 8, 1958,  
pp 1851-1859.

JPR8 2842

Sci - Chem

Jul 60

120,061

Enthalpy of Corundum in the Temperature Range From  
100 to 900° C, by K. Z. Gomelskiy, 8 pp.

RUSSLAN, per, Zhur Fiz Khim, Vol XXXIII, No 8, 1958,  
pp 1859-1862.

Dept of Commerce  
Nat'l Bur of Standards  
Dr Henry Birnbaum

RECORDED  
Sci - Chemistry  
Jun 59

89,476

APPROVED FOR RELEASE: 09/14/ 2001

CIA-RDP84-00581R000300910047-4

Study of Initiation and Propagation of Spherical Detonation, by V. B. Lazan.

REVIEWED, por, Phuc Phu Khim, Vol XXXI, No 6,  
1958, pp 1869-1872. 9072401

Entry 2012/T-58

Sci - Physics  
Feb 59

81,165

APPROVED FOR RELEASE: 09/14/ 2001

CIA-RDP84-00581R000300910047-4

Leont'ev, E. A. and Luk'yannovich, V. M.  
TECHNIQUE DE PREPARATION DES CORPS POREUX  
ET PULVERULENTS POUR L'ETUDE AU MICRO-  
SCOPE ELECTRONIQUE PAR LA METHODE DES  
REPLIQUES [Tekhnika Preparirovaniya pri Provodimii  
Elektronno-Mikroskopicheskikh Issledovaniy Poristykh  
i Peroshkoobraznykh i Metodom Replik] (Technique  
for Preparation of Porous Bodies and Powders for  
Electron Microscopic Study by the Replica Method) tr.  
by Bouchicot. 21 Jan 60 [14]p. 8 refs. CEA Trans.  
no. R 812 (text in F; each).  
Order from OTS or SLA \$1.60

61-23784

Trans., in French of Zhurnal Fizicheskoi Khimii  
(USSR)1958, v. 32, no. 8, p. 1922-1925

DESCRIPTORS: \*Porous materials, \*Powders, Electron

(Physics--Optics, TT, v. 6, no. 7)

61-23784

- I. Title: Replicas
- II. Leont'ev, E. A.
- III. Luk'yannovich, V. M.
- IV. CEA-tr-R812
- IV. Commissariat à  
l'Energie Atomique  
(France)

Office of Technical Services

Fedotov, N.A.  
PROPRIETES DES ELECTRODES DE VERRE AU  
LITHIUM (Elektrodyna Svoistva Litievogo Stekla) (The  
Electrode Properties of Lithium Glass), 19p. (Foreign  
text included 02refs. CNRS-VIII bis 487.  
Order from OTS, ETC or CNRS 50,80 - TT-62-28257

Trans. in French of Zhurnal Fizicheskoi Khimii (USSR)  
1958, v.32, no.9, p.1951-1957.

DESCRIPTORS: \*Glass electrodes, \*Lithium,  
Chemical properties, High temperature research,  
Measurement.

(Chemistry--Physical, TT, v. 11, no. 8)

TT-62-28257

I. Fedotov, N.A.  
II. CNRS-VIII bis 487  
III. Centre National de la Recherche Scientifique, Paris

Office of Technical Services  
European Translations Centre

Investigation of Hydrogen Over-Voltage on  
Germanium, by Ye. A. Yefimov, I. G. Yermolovich,  
9 pp.

RUSSIAN, per, Zhur Fiz Khim USSR, Vol XXXII,  
No 9, 1958, pp 1967-1970.

SLA 59-15004

Sci  
Dec 59  
Vol 2, No 4

103085

Bogoyavlenski, P. S.  
THE EFFECT OF LARGE MONOVALENT IONS ON  
THE STRUCTURE OF WATER AND ITS INFLUENCE  
ON THE SOLUBILITY OF SALTS. [1962] 8p.  
Order from ATS \$13.50                           ATS-OOP65R

Trans. of Zhurnal Fizicheskoi Khimii (USSR) 1958,  
v. 32, no. 9, p. 2035-2041.

DESCRIPTORS: \*Water, Molecular structure, Ions,  
\*Salts, Solubility.

(Chemistry--Physical, TT, v. 9, no. 5)

63-12474

I. Bogoyavlenski, P. S.  
II. ATS-OOP65R  
III. Associated Technical  
Services, Inc.,  
East Orange, N. J.

ATS R7-3742

Office of Technical Services

APPROVED FOR RELEASE: 09/14/ 2001

CIA-RDP84-00581R000300910047-4

On the Permeability of Iron in Hot Concentrated  
Alkaline Solutions, by A. M. Sukhotin, K. N.  
Kertchnev. UNCLASSIFIED

pp.

RUSSIAN, par, Zhur Fiz Khim, Vol XXXII, No 9, 1958,  
pp 2060-2072. 907677

Navy 2365/NRL 757

Sci - Physics  
Nov 59

CIA/TOD X-3756

101413

APPROVED FOR RELEASE: 09/14/ 2001

CIA-RDP84-00581R000300910047-4

Phase and Volume Correlations in Liquid-Gas Systems Under High Pressures. VI.. System Cyclohexane-Hydrogen, by I. P. Mrichevskiy, F. T. Sorina, 11 pp.

RUSIAN, per, Zhur Fiz Khim, Vol. XXXII,  
No 9, 1958, pp 2080-2086.

SLA 60-14057

Sci  
Vol IV, No 5  
Mar62

188, 446

<p>Verigin, N. N. DIFFUSION À LA SURFACE D'UN CORPS SOLIDE PLONGÉ DANS UN LIQUIDE [Diffuziya u Poverkhnosti Tverdogo Tela, Nakhodyashchegosya v Zhidkosti] [Diffusion at the Surface of a Solid Body Immersed in a Liquid] tr. by M. Grekow. 22 June 60 [22]p. 10 refs. CIA Trans. no. R 984 (text in French). Order from OTS or SLA \$2.60 61-15803  Trans. in French of <i>Zhurnal Fizicheskoy Khimii</i> (USSR) 1958, v. 32, no. 9, p. 2097-2106.</p> <p>DESCRIPTORS: *Diffusion, Solids, *Corrosion, Concrete, Rock, Liquids.</p> <p>(Chemistry--Physical, TT, v. 6, no. 2)</p>	<p>61-15803 I. Verigin, N. N. II. CEA-tr-R984 III. Commissariat à l'Énergie Atomique (France)</p> <p>10970</p> <p>Office of Technical Services</p>	
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Investigation of the Absorption of Ions on Platinized RR  
Platinum by Measuring the Adsorption Potentials, by  
A. D. Obrusheva, 13 pp.

RUSSIAN, part, Zvezda Khim, Vol XXII, No 9, 1956,  
pp 2155-2156.

NDF-O-010

Sci  
Nov 59

101, 331

Effect of Ionizing Radiation on Aqueous  
Solutions of Potassium Nitrate, by  
A. M. Kabakchi, V. A. Gramolin, 11 pp.  
RUSSIAN, per, Zhurnal fizicheskoiy khimiyy,  
Vol 32, 1958, No. 9, pp 2149-2154.  
AEC-RFP-TRANS P911263468-V

Sci/Chem  
Oct 68

369,446

Dissolution Anodique des Metaux Dans les Sels Fondus,  
II. Beryllium. (Anodic Dissolution of Metals in  
Fused Salts. II. Beryllium), by M. V. Smirnov,  
N. Ya. Tuchukreyev, 21 pp.

RUSSIAN to FRENCH, per, Zhur Fiz Khim, Vol XXXII,  
1958, pp 2165-2173.

Reverse Trans  
CEA tr R-653

Nov 59

Cathodic Processes in the Deposition of  
Boron-Lithium From Fused Electrolytes, by M. V.  
Smirnov, L. V. Ivanovskiy, 15 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, 1958,  
pp 2174-2182.

ABC FR-4416

Sci - Phys  
May 61.

150,330

Phenomenes Accompagnant le Depot Electrolytique du  
Beryllium a Partir des Sels Fondus. (Phenomena  
Accompanying the Electrolytic Deposition of  
Beryllium From Fused Salts), by M. V. Smirnov,  
L. E. Ivanovskiy, 18 pp.

RUSSIAN to FRENCH, per, Zhur Fiz Khim, Vol XXXII,  
1958, pp 2174-2182.

Reverse Trans  
CEA tr R-661

Nov 59

Determination of the Specific Surface Areas of  
Porous Bodies and of Powders by the Method of  
Gaseous Flow Under Low Pressure, by S. G. Entelin.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 9,  
1958, pp 2187-2191.

ATS RJ-2167

Sci - Chem

May 60

115,445

On the Nature of Cold Flame Phenomena, by N. S.  
Khikolopian, et al.

EN  
RUSSIAN, per, Zbir Pis Khim, Vol XXXII, No 9,  
1958, pp 2224-2228.

Navy 1989/T-59

Sci - Physics  
Jan 59

79, 884

Polymerization of Chlorinated Styrene Derivatives and Their Copolymerization with Vinyl Chloride, by G. V. Tkachenko, L. V. Stupen<sup>1</sup>, et al., 10 pp.

RUSSIAN, per, Zhur Fiz Khim, 1958, Vol XXXII,  
No 10, pp 2250-2255.

SLA 59-10445

Sci - Chem  
GMS I, 12  
Jul 59

92, 262

<p>Balashova, N. A. ADSORPTION DES IONS BROME ET IODE SUR LE PLATINE [Adsorbsiya Ionov Bromia i Yoda na Platine] [Adsorption of Bromine and Iodine Atoms on Platinum] tr. by Mine. Simon. 5 July 60 [21]p. 28 refs. CEA Trans. no. R 964 (text in French). Order from OTS or SLA \$2.60      61-23780  Trans. in French of Zhurnal Pizicheskoi Khimii (USSR) 1958, v. 32, no. 10, p. 2266-2273. A translation in English is available from MDR \$6.00 as MDP B-151 [1959] 12p.  DESCRIPTORS: *Platinum electrodes, *Adsorption, Bromine, Iodine, Ions.  For abstract see Technical Translations 3: 566, 1960.  (Chemistry--Physical, TT, v. 6, no. 6)</p>	<p>61-23780 I. Balashova, N. A. II. CEA-tr-R964 III. Commissariat à l'Énergie Atomique (France)  176685  Office of Technical Services</p>
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The Cause of the Anticorrosion Effect of Sodium Nitrite on Iron, by N. A. Shishakov, 6 pp.  
UNCLASSIFIED

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 10,  
1958, pp 2282-2286. 9676948

Navy 2366/DRL 758

Sci - Physics  
Nov 59

CIA/TDD X-3757

101415

Separation of the Photochemical From the Light-Filtering Effect and Latent Photosensitization, by Yu. S. Zuev.

RUSSIAN, per, Zhur Fiz Khim, Vol XXII, No 10, 1958,  
pp 2308-2314.

AT&T RJ-1724  
SLA 62-18171

Sci - Chem  
Jul 59

90, 458

The Conversion of Methane to Acetylene in the  
High Voltage Arc, Part I, by E. N. Yeremin, et al.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 10,  
1958, p 2315 (SIN 7211247)

2323

Navy 2232/T-147

ATS - 6 T - 1317

Acetone - Acet. 100%

TG 2000

SIA 60-21928

Scint Chem  
Jul 59

92, 812

The Promoting Effect of Cobalt Iron on the Chromate  
Ion Catalysed Decomposition of Hydrogen Peroxide, by  
G. L. Petrova, G. A. Bogdanov.

RUSIAN, per, Zhur Fiz Khim, Vol XXIII, No 10, 1958,  
pp 2324-2332.

ILL RPS 2269 (200-1)

Sci - NM  
Feb 64

248, 283

The Kinetics and Mechanism of Catalytic Conversions  
of Acetylene, II, Some Questions of the Mechanism  
of the Addition of Various Molecules to Acetylene, by  
R. M. Flid, 15 pp.

RUSSIAN, per., Zhur Fiz Khim, Vol XXXII, No 10,  
1958, pp 2339-2346.

SLA 59-10444  
AT&T - RJ-1908

97,281

Sci - Chem  
Sep 59  
Vol 2, No 1

Vagin, E. V. and Zhukhovitskii, A. A.  
THEORY OF THERMAL-ADSORPTIVE SEPARATION.  
26 June 61, 25p. 6 refs.  
Order from OTS or SLA \$2.60

63-10216

Trans. of Zhurnal Fizicheskoi Khimii (USSR) 1958,  
v. 32, no. 10, p. 2362-2373.

DESCRIPTORS: \*Adsorption, Separation, Gases, Sorption, Chromatographic analysis, Thermal diffusion, Equations of state, Theory.

A theory of thermal-adsorptive separation of gases based on the assumption of pressure equality in the sorbent layer is presented. In the case of an infinitely great sorption rate and of absence of longitudinal diffusion in the process of thermal desorption, a distribution of the adsorbed substance corresponding to the temperature field is established, the motion velocities (Chemistry--Physical, TT, v. 10, no. 3) (over)

63-10216

I. Vagin, E. V.  
II. Zhukhovitskii, A. A.

Offices of Technical Services

On the Charging of a Semiconductor Surface  
During Adsorption, by F. F. Vol'kenshteyn, 12 p.

RUSSIAN, per, Zhur Fiz Khim, 1958, Vol XXXII,  
No 10, pp 2383-2391.

MDF V-195

Sci  
Mar 60  
Vol 2, No 11

109,670

On the Nature of the Negative Temperature Coefficient of the Rate of Radical Combination, by A. D. Stepankovich.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 10, 1958, p #415. -2417.

Sci - Physics  
Jul 59

Navy 2233/T-148  
OTS 60-21787

92, 813

The Radiation Oxidation of Nitrogen. III. Some Questions Concerning the Mechanism of the Reaction and a Comparison With Electrical Discharge Data, by N. T. Dmitryev.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 10,  
1958, p 2418. -2423

Sci - Physics  
Jul 59

Navy 2423/T-149  
OTS 60-21784

92, 815

The Present State of Research on the Phase Diagram  
of the Boron-Carbon System, by G. V. Samsonov, 7 pp

RUSSIAN, per, Zhur Fiz Khim USSR, Vol XXXII, No 10,  
1958, pp 2424-2429.

R  
ATS-55L31W

Sci  
Dec 59  
Vol 2, No 4

102, 778

Electrochemical Properties of Zirconyl  
Phosphate. Part 1. Electrode Properties  
of Membranes, by E. A. Materova.  
RUSSIAN, per, Zhurnal Fizicheskoi Khimii,  
Vol 38, 1964, pp 676-680.  
NTC 72-10454-07D

Apr 72

The Rate of Desorption, by D. P. Timofeyev.  
RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 11,  
1958, pp 2483-2486.

NLL M. 2577

Sci - Chem

173, 878

Nov 61

<p>Tkachenko, G. V., Stupen, L. V. and others. THE COPOLYMERIZATION OF VINYL CHLORIDE WITH METHACRYLIC ACID ESTERS. [1961] [9p]. 10 refs. Order from OTS or SLA \$1.10                  61-20705</p> <p>Trans. of Zhurnal Fizicheskoi Khimii (USSR) 1958, v. 32, no. 11 [p. 2492-2499].</p> <p>DESCRIPTORS: *Vinyl chlorides, Copolymerization, Polymerization, *Acrylic acids, Acids, Esters, Benzozates, Polymers, Molecular weight.</p> <p>The copolymerization of vinyl chloride with methyl-, n-butyl-, n-octyl methacrylates and vinyl benzoate was studied. The rates of the copolymerization of vinyl chloride with methacrylates and the molecular weights of the polymers in all of the corresponding monomers are less than with acrylates. These values (Chemistry--Organic, TT, v. 7, no. 10)                  (over)</p>	<p>61-20705</p> <p>I. Tkachenko, G. V. II. Stupen, L. V.</p> <p>Office of Technical Services</p>
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Yeremin, Ye. N.  
THE KINETICS OF ELECTROCRACKING OF METHANE [K Voprosu o Kinetike Elektrokrekinga Metana].  
[1961] 7p.  
Order from ATS \$12.55

ATS-52NS1R

Trans. of Zhur[nal] Fiz[icheskoy] Khim[ii] (USSR)  
1958, v. 32, no. 11, p. 2543-2549.

DESCRIPTORS: \*Methanes, Decomposition, Hydrocarbons, Reaction kinetics.

(Chemistry--Physical, TT, v. 6, no. 3)

61-22612  
763-18473  
I. Yeremin, Ye. N.  
II. ATS-52NS1R  
III. Associated Technical Services, Inc., East Orange, N. J.

ATS-RJ-2892

76323

Office of Technical Services

Studies in the Mechano-Chemistry of Polymers  
XXIII. Mechanical and Chemical Processes  
in the Mastication of PVC, by A. A. Berlin,  
et al.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII,  
No 11, 1958, pp 2565-2570.

CSIRO

207, 417

Sci - Chem  
Aug 62

The Use of the Electric Arc in Calorimetry,  
by A. F. Vorob'yev, S. M. Skuratov, 8 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 11,  
1958, pp 2580-2585.

SLA 60-13082

Sci  
Aug 61

OIS, Vol III, No 12

163,132

S-65/60

(NY-3554).

On the Computation of a Single Coefficient of the Separation of Isotopes for Equilibrium Processes,  
by Ye. M. Kuznetsova, A. V. Makarov, G. M.  
Panchenkov, 3 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 11,  
1958, pp 2641-2643.

JPRS 2476

114 730

Sci - Chem  
Apr 60

1/19/59

"The 40 th Anniversary of the Physico-Chemical  
Institute imeni L. Ya. Karpov" /unsigned article/,  
Zhurnal Fizicheskoy Khimii, Vol 32, No 12,  
Dec 58, pp 2657-2662.

SIR T-16.

NCH-202 507      Field 7D

Sinytsina, Z. A.  
THE INHIBITING EFFECT AND THE STRUCTURE OF  
INHIBITORS. 2. THE INHIBITED POLYMERIZATION  
OF METHYLACRYLATE. 1965, 16p. TC-1221.  
Order from TC: \$15.00 as TC-1221

Trans. of Zhurnal Fizicheskoy Khimii (USSR) v32 n12  
p2663-73 1958.

I. Trans-Chem  
Knoxville, Tenn.  
II. Title: Inhibited...

The Electrochemical and Corrosion Behavior of  
Steel and Nickel Electrodes in Sulphuric Acid  
Solutions Subjected to the Action of Rays, by  
N. Ya. Bune, Ya. M. Kolotyrkin, G. S. Tyurikov,  
11 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, 1958,  
pp 2679-2685.

AEC NP Tr-445

Sci - Chem

125, 848

Relation Between the Energy Yield and the Kinetics  
of a Radiation-Chemical Reaction, by S. Ya.  
Psheshetskiy, M. T. Dmitryev. UNCLASSIFIED

RUSSIAN, per, Zhur Fiz Khim, Vol XXII, No 12,  
1958, pp 2680-2689. 7C78.47

Navy 2211/MRL 727

Sci - Physics, Chem  
Jul 59

92, 828

Shatenshtain, A. I. and Izrailevich, E. A.  
COMPARISON OF DEUTERO EXCHANGE AND  
METALATION OF ALKYL BENZENES. [1962] [9]p.  
30 refs.

Order from OTS or SLA \$1.10      63-15923

Trans. of Zurnal Fizicheskoi Khimii (USSR) 1958  
[v. 32] no. 12, p. 2711-2716.

DESCRIPTORS: \*Metalorganic compounds, \*Potassium  
compounds, Amides, \*Alkyl radicals, \*Benzenes,  
\*Deuterated compounds, Exchange reactions, \*Metal-  
ation, Reaction kinetics

The rate of deutero exchange between a solution of  
potassiumamide (0.02N) in liquid deuterioammonia and the  
aromatic rings and alkyl groups of the following alkyl  
benzenes: Q<sub>1</sub>H<sub>5</sub>ClH<sub>3</sub>, iso-Q<sub>2</sub>H<sub>5</sub>C<sub>3</sub>H<sub>7</sub>, tert-C<sub>6</sub>H<sub>5</sub>C<sub>4</sub>H<sub>9</sub>,  
and n-C<sub>6</sub>H<sub>5</sub>C<sub>4</sub>H<sub>9</sub> has been measured at 25, 10, and 0°C.  
(Chemistry--Physical, TT, v. 10, no. 1)      (over)

63-15923

I. Shatenshtain, A. I.  
E. Izrailevich, E. A.  
III. Translations, New York

Office of Technical Services

The Use of Radioactive Sulfur to Study  $H_2S$ -Poisoning of Ammonia-Synthesis Catalyst, by  
Yu. I. Buletnikova, L. O. Apel'bayev, N. I.  
Zemtsev.

RUDNIT, per, zhur fiz anim, Vol 40(1), No 12,  
1958, pp 2717-2724.

DSIR LLW M.791(Con) Sci. + Chem  
Jan 60

Dept of Interior  
TN7, E57, No 162  
SLA 59-18543 105046

Dissociation of Deuteroctane Under Electron Impact, by M. V. Gur'yev, M. V. Nikhamirov, 10 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII,  
No 10/12, 1958, pp 2731-2738.

AT&T-08M42R

AT&T/RJ-2545

Sci.

Vol IV, No 11  
June 62

199, 250

Boreskov, G. K.  
CATALYST INTERACTION WITH THE REACTION  
SYSTEM. [1963] 10p. 20 refs.  
Order from OTS or SLA \$1.10

63-14330

Trans. of Zhur[nal].R[uskae]e[boekot] Khim[ic]h (USSR) 1958,  
v. 32 [no. 12] p. 2739-2747.

DESCRIPTORS: \*Catalysts, Chemical reactions, Theory,  
Metals, Oxides, Reaction kinetics.

In the course of a catalytic reaction, besides the  
action of the catalyst on the reagents, there is always  
an action on the reacting system on the catalyst  
leading to changes in its composition and consequently  
in its properties. To each composition of the reacting  
mixture there corresponds a definite, steady state  
composition of the catalyst depending upon the ratio of  
(Chemistry-Physical, TT, v. 10, no. 3) (over)

63-14330

I. Boreskov, G. K.

Office of Technical Services

The Combustion of Methane to Acetylene in a High-Voltage Arc. Part 2. The Effect of Hydrogen, by E. N. Kremin,

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 12, 1958,  
pp 2767-2771.

Sci - Chem

AT&T RJ-2073

AT&T-256137K

May 60

115379

The Mechanism of Formation of H<sub>2</sub>O and H<sub>2</sub>O<sub>2</sub> in the  
Reaction of Hydrogen Atoms With Oxygen Molecules,  
By L. I. Avramenko, R. V. Molesnikova. 1970.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 12,  
1958, pp 2780-2786.

DSIR LIU RTS 1218

(17a. 6a.)

Sci - Chem  
Nov 59

100,787

The Kinetics of Consecutive Reactions in a Recycle-Flow System, by S. V. Dobrovol'skiy, V. Yu. Polotnyuk.

RUSSIAN, per, Zhur Fiz Khim, VOL XXXII, NO 12, 1958, pp 2792-2795.

Sci - Chem  
Dec 59  
*103 771*

SLA 61-17377  
AT&T RJ-2037

Bochnar, D. A., Gambaryan, N. P., Stankevich, I. V.,  
and Chistyakov, A. L.

A QUALITATIVE DETERMINATION OF THE STA-  
BILITY OF HETERO CYCLIC SYSTEMS WITHIN THE  
FRAMEWORK OF THE HÜCKEL APPROXIMATION.  
[1963] 9p. 2refs.

Order from OTS, SLA, or ETC \$1.10 TT-63-20684

Trans. of Zhur[nal] Fiz[icheskoi] Khim[iii] (USSR) 1958,  
v. 32 [no. 12] p. 2797-2802. (Abstract available)

DESCRIPTIONS: \*Heterocyclic compounds, \*Organoboranes, Chemical bonds, Stability, Atomic orbitals, Polynomials,

A purely qualitative method is proposed for deciding, within the framework of the Hückel approximation, the question of the presence of a "closed electron shell" in conjugated systems and, in this sense, the question of the relative stability of such systems. This method con-  
(Physics--Molecular, TT, v. 11, no. 3) (over)

TT-63-20684

1. Title: Borepin
  2. Title: Hückel approximation
  3. Title: Molecular orbital theory
- I. Bochnar, D. A.
  - II. Gambaryan, N. P.
  - III. Stankevich, I. V.
  - IV. Chistyakov, A. L.

Office of Technical Services

Thermochimistry of Isotopes. I. I. Calorimetric  
Investigation of the Heat of Formation of Hydrogen  
Sulfide and Deuterium Sulfide, by A. P. Kayestinskii,  
R. T. Kan'kovskiy, 12 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXII, No 12, 1958,  
pp 2610-2616.

US Dept of Commerce  
National Bur of Standards

Sci - Phys  
May 61

148,776L

Filippov, Yu. V. and Yemel'yanov, Yu. M.  
ELECTRICAL THEORY OF OZONIZERS. III. THE  
ELECTRIC CURRENT IN OZONIZERS (Elektriches-  
kaya Teoriya Ozonatorov. III. Elektricheskiy Tok v  
Ozonatorakh) tr. by E. Feigl. [1960] [14]p. (foreign  
text included) 5 refs. Trans. no. 4850.  
Order from LC or SLA m\$2.40. ph\$3.30 61-15173

Trans. of Zhurnal Fizicheskoy Khimii (USSR) 1958,  
v. 32, no. 12, p. 2817-2823.

The previous theory (Zhur. Fiz. Khim. 31: 1628, 1957)  
is used to derive expressions for the dependence of the  
effective and mean values of the strength of the current  
flowing through the ozonizer on (1) the parameters of  
the apparatus and (2) the magnitudes of the voltage ap-  
plied to the terminals. (See also 59-10012)

(Engineering--Electrical, TT, v. 5, no. 5)

61-15173

1. Electric discharges
2. Ozone--Production
3. Title: Ozonizer
- I. Filippov, Yu. V.
- II. Yemel'yanov, Yu. M.
- III. Title: Electric...
- IV. CSIRO Trans-4850
- V. Commonwealth Scientific  
and Industrial Research  
Organization (Australia)

146,964

Office of Technical Services

12 216

US-14 \* UK-12

GUR'EV M. V., TIKHOMIROV M. V., TUNITSKII N. N.

The mass spectra of large molecules

\* Zhur. Fiz. Khim., 32, No. 12, p. 2847 (1958)

US-14: ATS-09M42R. \$ 2.75 - English

\* UK-12: English

E u r a t o m  
ATS/RJ-2546

The Synthesis of Some Deuterated Aromatic Hydrocarbons and their Raman Spectra, by A. I. Shatenshtain.

RUSSIAN, par., Zh Fiz Khim, Vol 32, No 1, 1958,  
pp 146-151.

HEM-NIH  
12-4-67

Sci/B & M  
Feb 68

319,425

ZHURNAL FIZICHESKOJ KHEMII -- monthly  
(Soviet Journal of Physical Chemistry)  
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*Vol 33, Nos 7-12, 1959*

*Cover-to-cover Cleaver-Hume Press LTD*

*Available Br 5 Library*

*No 7, pp 1674-1675 JPRS 3368*

*No 12, pp 2755-2766 JPRS 2736*

Viscosity Measurements with Microamounts of  
Liquid, by V. S. Bondarenko.  
RUSSIAN, par, Zhurnal Fizicheskoi Khimii, Vol 38,  
1964, pp 1273-1275.  
NTC 71-14363-07D

Mar 72

(NY-4844)

CO + O<sub>2</sub> Flame Radiation and the Mechanism of This  
Reaction, by M. Zateiorskiy, V. Kondrat'ev, S.  
Solnyshkov, 14 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XIV, No 12, 1960,  
pp 1521-1527.

JPRS 7532

Sci - Chem

Feb 61.

139.356

Electrosynthesis of O<sub>3</sub>. I. Kinetics of the  
Synthesis of O<sub>3</sub> Under Flow Conditions, by  
Yu. V. Filippov, Yu. N. Emol'yanov.  
RUSSIAN, par., Zh Fiz Khim, Vol 35, 1961,  
pp 407-415.  
AEC-ANL-Trans-522

Sci-Chem  
Jan 68

348,751

Zarif'yants, Y. A., Kiselyev, V. F., and  
Fedorov, G. G.

ADSORPTION CHIMIQUE DE L'OXYGÈNE A LA SURFACE D'UN GRAPHITE. ÉCHAUFFEMENT FORMÉ(?) A  
cheskaya Adsorbsiya Kisloroda na Poverkhnosti Svezhego Raskola Grafita) tr. by Mihailovic. 26 Dec 61  
[8]p. (Russian text included) 3 refs. CEA Trans.  
no. R 1452 (text in French).

Order from OTS or SLA \$1.10 62-24573

Trans. in French of \*Zhurnal] Fiz[icheskoi] Khim[ii]  
(USSR) 1961, v. 35, no. 8, p. 1885-1886.

DESCRIPTORS: \*Adsorption, \*Oxygen, \*Graphite,  
Cooling, Temperature.

(Chemistry--Physical, TT, v. 8, no. 3)

62-24573

I. Zarif'yants, Y. A.  
II. Kiselyev, V. F.  
III. Fedorov, G. G.  
IV. CEA-tr-R1462  
V. Commissariat à l'Energie  
Atomique (France)

Office of Technical Services

(SF-1878)

Ukrainian Republican Conference on Adsorption and Methods of Chromatographic Analysis, by K. V. Chmutov, 6 pp.

RUSSIAN, per, Zhur Fiz Khim, Vol XXXV,  
No 9, pp 2170-2172.

JPRS 13<sup>6</sup>70

Sci. - Chem

May 62

*193, 511*  
*Keep in file*

(SF-2212)

M. M Dubinin, by B. P. Bering, V. V. Serpinskiy,  
5 pp.

RUSSIAN, per, Zhurnal Fizicheskoy Khimii,  
No 11, 1961, pp 225-227.

JPRS 13882

USSR  
Biog  
Jun 62

Kobozev, N. I., Semiolokin, I. A., and Pitskhelauri, E. N.

REACTION BETWEEN HYDROGEN AND OXYGEN IN THE SILENT ELECTRICAL DISCHARGE. VIII. MECHANISM OF THE PROCESS (Vzaimodeistviye Vodoroda s Vodorodom v Tikhom Elektricheskem Razryade).

[1962] [15]p. 14 refs.

Order from UTS or SLA \$1.60

63-10748

Trans. of Fizicheskoy Khimii (USSR) 1962, v. 36 [no. 3] p. 443-448.

DESCRIPTORS: \*Hydrogen, \*Oxygen, \*Electric discharges, Chemical reactions, \*Reaction kinetics, Electrochemistry.

Water vapors and particularly argon have been shown to be energy catalysts of the formation and of the decomposition of  $H_2O_2$ . At low concentrations they act (Chemistry--Physical, TT, v. 10, no. 3) (over)

63-10748

I. Kobozev, N. I.  
II. Semiolokin, I. A.  
III. Pitskhelauri, E. N.  
IV. Title: Mechanism ...

Office of Technical Services

Vysotskite-a New Sulphide of Palladium and Nickel, by A. D. DGenkin, O. E. Zvyagintsev.  
RUSSIAN, per, Zapiski Vsesoyuznogo Mineralogicheskogo Obshchestva, Vol 91, No 6, 1962,  
pp 718-725.  
NLL RTS 2713

Sci - Earth Sci & Ocean  
Sept. 67

340,391

Marina, L. I., Nashel'ski, A. Ya., and Yakobson,  
S. V.  
INVESTIGATION OF THE DECOMPOSITION PRESSURE  
OF GALLIUM PHOSPHIDE. [1962] 11p. 16 refs.  
Order from OTS or SLA \$1.60

63-10772

Trans. of #Zhurnal Fizicheskoi Khimii (USSR) 1962,  
v. 36, no. 5, p. 1086-1088.  
Another trans. is available from OTS or SLA \$1.10  
as 62-18968, 2 Aug 62 [8pp].

DESCRIPTORS: \*Semiconductors, \*Gallium compounds,  
\*Phosphides, Decomposition, Pressure, Crystallization,  
Temperature, Phosphorus, Vapor pressure.

The temperature dependence of the decomposition  
pressure of gallium phosphide expressed in the following  
terms has been determined

$$\lg P = - \frac{10770}{T} + 9.996.$$

(Chemistry-Physical, TT, v. 9, no. 11) (over)

63-10772

I. Marina, L. I.  
II. Nashel'ski, A. Ya.  
III. Yakobson, S. V.

Office of Technical Services

Electrosynthesis of Ozone. V. Synthesis of Ozone  
in a Flow Under Reduced Pressure, by V. G. Samoilovich,  
V. P. Vendillo.

RUSSIAN, par, Zhur Fiz Khim, Vol XXXVI, No 5, 1962,  
pp 989-992.

WPTD-TT-62-1462

Sci - Chem

Oct 62